

Single-step Synthesis of Coenzyme Q₀

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Authors' contributions

This work was carried out in collaboration among all authors. Author JW designed the study. Author YYY performed the statistical analysis. Author YFQ wrote the protocol. Author TLZ wrote the first draft of the manuscript. Authors YBH and SQ managed the analyses of the study. Authors WYL and YZ managed the literature searches. All authors read and approved the final manuscript.

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ABSTRACT

A new method for the preparation of 2-methyl-5,6-dimethoxy-1,4-benzoquinone (Coenzyme Q₀) was developed. This improved process in one step by the oxidation of 3,4,5-trimethoxytoluene to coenzyme Q₀ by simple oxidation using potassium or ammonium persulfate under transition-metal free conditions.

Keywords: Coenzyme Q₀; 3,4,5-trimethoxytoluene; potassium persulfate.

1. INTRODUCTION

Coenzyme Q₁₀ (CoQ₁₀, Fig. 1), also known as ubiquinone, is a vitamin-like 1,4-benzoquinone compound [1] and functions as a potent

antioxidant that scavenges free radicals [2]. CoQ₁₀ is widely used in the treatment of cardiovascular disease, mitochondrial disorders [3], and in the improvement of immunotherapy [4]. 2,3-Dimethoxy-5-methyl-1,4-benzoquinone,

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known as Coenzyme Q₀ (CoQ₀, Fig. 1), is a key constituent part of coenzyme Q₁₀. Coenzyme Q₀ has been reported possess antineoplastic, anti-inflammatory and antimicrobial activities [5].

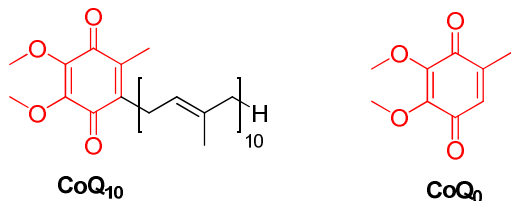


Fig. 1. Structures of CoQ₁₀ and CoQ₀

There have been several methods published for the preparation of Coenzyme Q₀ through oxidation of commercially available 3,4,5-trimethoxytoluene (1) with the oxidant-hydrogen peroxide (H₂O₂) system. Among metal catalysts applied were potassiumhexacyanoferrate(III) K₃Fe(CN)₆ [6], methyltrioxorhenium (CH₃ReO₃) [7], ruthenium complex-bound norvaline [8] and γ-Keggin divanadium-substituted phosphotungstate [9]. Recently, Bjørsvik et al. utilized hydrogen peroxide in combination with mineral acids (HNO₃) [10] to produce CoQ₀, which imposed practical problems related to reactor corrosion and safety risks. Based on our previous study [11], here we described a single step synthesis of CoQ₀ by treatment of 3,4,5-Trimethoxytoluene 1 with persulfate (K₂S₂O₈, Na₂S₂O₈, (NH₄)₂S₂O₈) under transition metal-free conditions (Table 1).

2. EXPERIMENTAL SECTION

All reactions were monitored by TLC (SiO₂, petrol ether/EtOAc 5:1), Melting points were measured on Melting Point M-565 (BUCHI). NMR and mass spectra were recorded on a Bruker Avanc III-HD 400 NMR and a TripleTOF Mass spectrometers, respectively. All reagents: e.g. Potassium Persulfate (K₂S₂O₈), Ammonium persulphate ((NH₄)₂S₂O₈), acetic acid were purchased from Adamas, P. R. China, and used without further purification.

General method for preparation of CoQ₀

3,4,5-Trimethoxytoluene (1.82 g, 10 mmol) was dissolved in a mixture of acetic acid (99%, 10 mL) and catalytic H₂SO₄ (0.1 mL), then a solution of oxidant (15 mmol) was added dropwise over 10 minutes. The mixture was stirred and heated at 50°C for 1 hour and extracted with CH₂Cl₂ (3 x 10 mL). The combined organic phases were washed with H₂O and saturated NaHCO₃, then dried over anhydrous Na₂SO₄, and evaporated under reduced pressure. The residue was purified by a silica-gel column chromatography (PE/EtOAc 5:1) to give coenzyme Q₀.

Coenzyme Q₀, red-colored needles, m.p. 55-58°C (Lit [12] 57-59°C).

IR (KBr/cm⁻¹): 3590, 3415, 1661, 1603, 1291, 1226, 999.

¹H NMR (400 MHz, CDCl₃) δ 6.44 (q, *J* = 1.7 Hz, 1H), 4.02 (s, 3H, OCH₃), 4.00 (s, 3H, OCH₃), 2.04 (d, *J* = 1.6 Hz, 3H, CH₃).

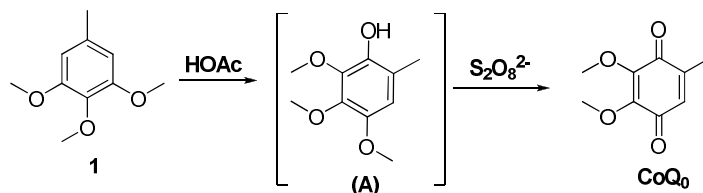
¹³C NMR (101 MHz, CDCl₃) δ 184.4 (C=O), 184.2(C=O), 145.0, 144.8, 144.0, 131.2, 61.2 (OCH₃), 61.1 (OCH₃), 15.4 (CH₃).

MS (ESI): *m/z* = 205 [M+Na]⁺.

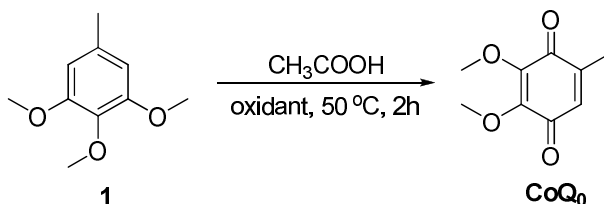
3. RESULTS AND DISCUSSION

As shown in Table 1, the reaction is conducted in acetic acid at 50°C in less than 2 h and without using any metal catalyst. The traditional method employing 30% H₂O₂ as oxidant give a yield of 50% (entry 1, Table 1). The use of Na₂S₂O₈ and (NH₄)₂S₂O₈ can improve the reaction yield (entry 3-4, Table 1). The best yield was obtained using K₂S₂O₈ as oxidant to afford the desired product CoQ₀ in 80% yield (entry 2, Table 1). Persulfate salts were first employed as oxidants instead of transition metal complexes as the catalyst to synthesize 1,4-benzoquinone under mild conditions, this chemistry is clean and easy to work up.

Based on the experiment results, a proposed reaction mechanism was illustrated in Scheme 1. Initially, compound 1 was transformed to intermediate (A) in the presence of HOAc solvent, then A was oxidated by S₂O₈²⁻ to give product CoQ₀ in situ.



Scheme 1. Proposed reaction mechanism

Table 1. Single-step synthesis of CoQ₀

Entry	oxidant	Solvent	Temp (°C)	Yield (%)
1	30% H ₂ O ₂	CH ₃ COOH	50	50
2	K ₂ S ₂ O ₈	CH ₃ COOH	50	80
3	(NH ₄) ₂ S ₂ O ₈	CH ₃ COOH	50	70
4	Na ₂ S ₂ O ₈	CH ₃ COOH	50	60

Reaction Conditions: compound 1 (0.01mol), oxidant (1.5 equiv), 2 hour under open air

4. CONCLUSION

In summary we have developed a high-yielding and selective synthetic protocols for the preparation of 2,3-dimethoxy-5-methyl-[1,4]benzoquinone (Coenzyme Q₀) from the cheap and readily available 3,4,5-Trimethoxytoluene **1** by oxidation using potassium persulfate in the presence of catalytic sulphuric acid. The reaction is efficient, clean and easy work-up. This method could be used for the synthesis of other coenzyme Q compounds.

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COMPETING INTERESTS

Authors have declared that no competing interests exist.

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